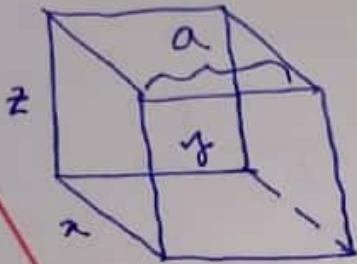


TOPIC $\frac{0}{0}$ TRANSLATIONAL PARTITION FUNCTION (MONOATOMIC GAS)

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Energy of a particle in 3-D box



$$E_{n_x, n_y, n_z} = \frac{h^2}{8ma^2} (n_x^2 + n_y^2 + n_z^2)$$

where $n_x, n_y, n_z = 1, 2, 3, \dots$

$$\Rightarrow Z_t = q_{\text{trans}} = \sum_{n_x=1}^{\infty} \sum_{n_y=1}^{\infty} \sum_{n_z=1}^{\infty} e^{-\beta E_{n_x, n_y, n_z}}$$

$$= \left(\sum_{n_x=1}^{\infty} e^{-\frac{\beta h^2 n_x^2}{8ma^2}} \right) \times \left(\sum_{n_y=1}^{\infty} e^{-\frac{\beta h^2 n_y^2}{8ma^2}} \right) \times$$

$$\left(\sum_{n_z=1}^{\infty} e^{-\frac{\beta h^2 n_z^2}{8ma^2}} \right)$$

$$= \left(\sum_{n=1}^{\infty} e^{-\frac{\beta h^2 n^2}{8ma^2}} \right)^3 = q_1^3$$

q_1 cannot be evaluated in closed form.
This discretized summation can be expressed in integral form.

$$q_1 = \int_0^{\infty} e^{-\left(\beta \hbar^2 n^2 / 8ma^2\right)} dn$$

Using standard integral

$$\int_0^{\infty} e^{-\alpha x^2} dx = \frac{1}{2} \left(\frac{\pi}{\alpha}\right)^{\frac{1}{2}}$$

$$q_1 = \frac{1}{2} \left(\frac{\pi}{\left(\beta \hbar^2 / 8ma^2\right)}\right)^{\frac{1}{2}}$$

$$q_1 = \left(\frac{2\pi m k_B T}{\hbar^2}\right)^{\frac{1}{2}} a$$

$$\rightarrow) \quad q_{\text{trans}} = q_1^3 = \left(\frac{2\pi m k_B T}{\hbar^2}\right)^{\frac{3}{2}} a^3$$

$$= \left(\frac{2\pi m k_B T}{\hbar^2}\right)^{\frac{3}{2}} V \quad \text{where } V = a^3 \\ = \text{volume}$$

If we consider rectangular parallelepiped

$$q_{\text{trans}} = \left(\frac{2\pi m k_B T}{\hbar^2}\right)^{\frac{1}{2}} a \left(\frac{2\pi m k_B T}{\hbar^2}\right)^{\frac{1}{2}} b$$

$$\times \left(\frac{2\pi m k_B T}{\hbar^2}\right)^{\frac{1}{2}} c$$

$$q_{\text{trans}} = \left(\frac{2\pi m k_B T}{\hbar^2}\right)^{\frac{3}{2}} V \quad \text{where, } V = abc \\ = \text{volume}$$

The same result.